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OBSERVATIONS OF COMETS: RADIO AND OTHERWISE

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THEORETICAL ASPECTS OF COMETARY RADIO LINE EXCITATION: OH, H2O AND OTHERS

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Models of radio and infrared molecular emissions in comets must account for the various processes listed in the first illustration (Fig. 7-1). The solar UV is the principle destructive agent for most cometary molecules via photodissociation, thus regulating the size of the emission region. The well-known Swings effect in the OH-molecules is also governed by the fine structure of the solar UV spectrum. Electronic transitions are less frequent for other important molecules (e.g. CO, H_2O), which are more easily excited to higher vibrational states by the solar infrared radiation. Collisional excitation and secondary radiation sources (nucleus and dust grains) need to be included in models of the inner coma. It has become apparent over the past few years that: (a) LTE conditions are invalid and cannot explain the distribution of rotational energy levels, and (b) specific calculations must be performed for each individual species ("simplest" assumption: fluorescence equilibrium).

Since each molecular type more or less "does its own thing", it is appropriate to comment briefly on some of the more widely-sought constituents of cometary comae:

OH: The excitation and fluorescence of the hydroxyl maser in comets is one of the very few processes in this field that could be classified as well known. Some minor discrepancies from the refined theory (Depois et al., 1981), such as the (as yet unconfirmed) post-perihelion emission predicted by Schleicher and A'Hearn (1982), still need to be resolved.

CO: Carbon monoxide, having a relatively long photodissociation lifetime, is expected to be in a state of fluorescence equilibrium in the outer coma with detectable emissions in the infrared (1-O band) and microwave (mostly submillimeter) ranges (Crovisier and Le Bourlot, 1983). Typical excitation rates (s⁻¹) at 1 AU and the expected relative populations of the rotational levels of CO are displayed in Fig. 7-2.

 H_2O : The water molecule is expected to show profuse fluorescence emission at IR and submillimeter wavelengths which cannot be observed at ground level due to absorption by telluric water. The radio line at $\lambda 1.35$ -cm cannot be explained by fluorescence models (Crovisier, 1984). A recent maser model by Strelnitskii (1983) is outlined in Fig. 7-3. The model has a compact emission region some 100 km in diameter which can account for the line width (thermal) and blue shift (optically thick) of the line observed in Comet Bradfield (1974 III). It fails to explain the detection in Comet IRAS-Araki-Alcock.

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m NH}_3$: Ammonia is a more fragile molecule in the solar radiation field (level diagram: Fig. 7-4), so its coma size may not be too much larger than the collision-dominated regime about the cometary nucleus (r = 5-8 x 10^3 km). Near-Earth approaches are thus quite favorable for its detection. The molecule is expected to assume primarily the metastable (J=K) states, so that a determination of the relative intensities of the K-band inversion lines could be used to derive the kinetic temperature of the coma.

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Modelling radio and IR molecular emission in comets.

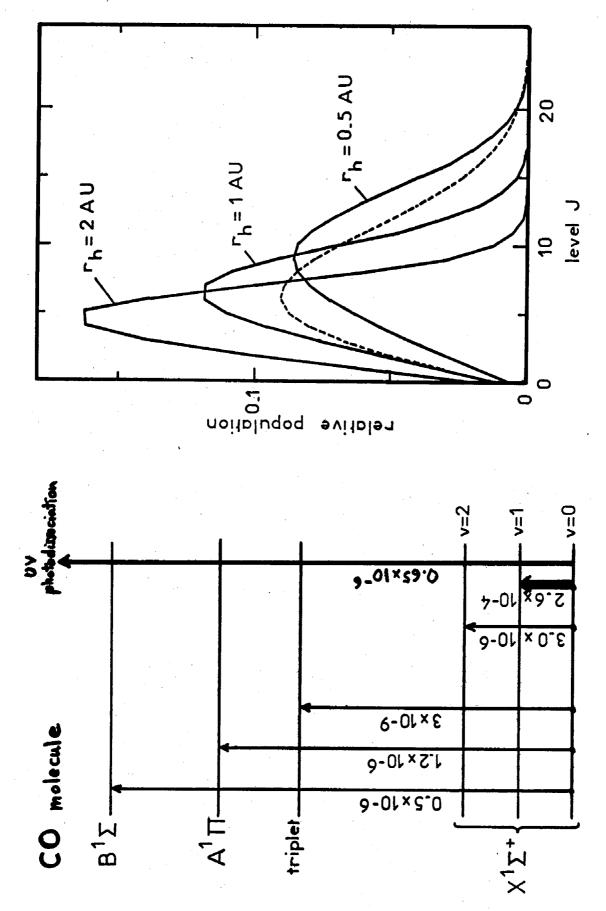
Excitation mechanisms -

- Solar UY

 mainly photo dissociation (10-3 10-65-1)
- -Solar IR

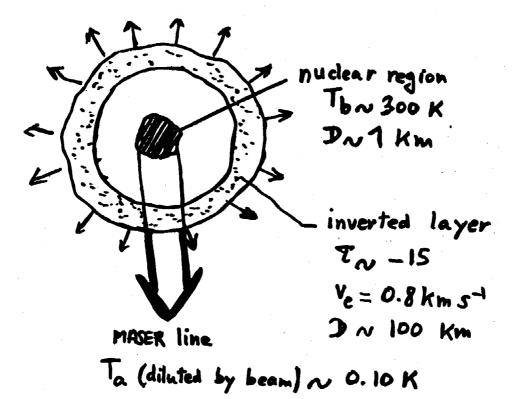
 prevailing mechanism-excites fundamental bands of vibration- g ~ 10-45-1.
- cometary dust IR in the inner come dust models needed.
- collisions
 in the inner coma- collision cross-sections neededtemperature law needed = very cold (~ 15K)
 at r ~ 50 100 km !
- Dynamical evolution of the rotational population distribution:

 _from thermal distribution (nucleus region)
 - to fluorescence equilibrium (outer coma)
- thermal emission (200K) not valid.
- > Specific calculations necessary for each molecule.
 - Fig. 7-1. Some aspects to be considered in modelling radio and infrared line emission from comets



Schematic excitation diagram (left) and relative population of rotational energy levels (right) for cometary CO (from Crovisier and Le Bourlot, 1983). Fig. 7-2.

H2O maser models for comets

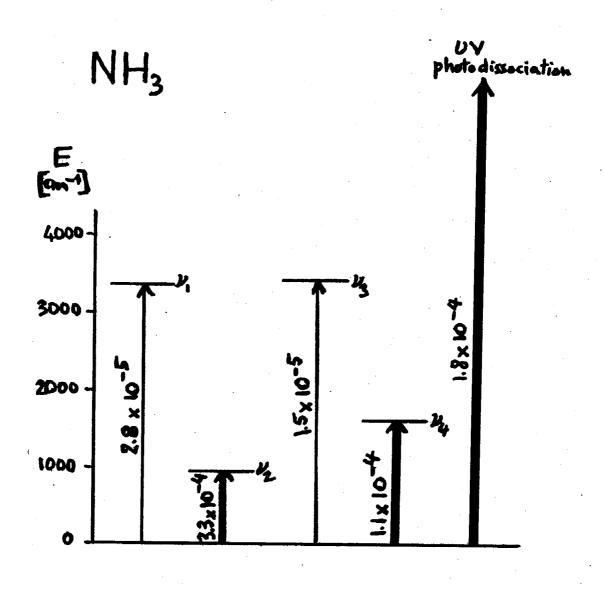


[from Strol nitskii, 1983]

-asknot

- explains Bradfield 1974 111.
- -does not explain I.-A.-A. 1983d-
- requires high Q(140) = 1029-1030 mobe. 5-1-
- the pumping mechanism must be figured out.

Fig. 7-3. A maser model for cometary H_2O (Strelnitskii, 1983)



life time
$$n.5.5 \times 10^3 \text{ s}$$

Aij (vib). $n.10 \text{ s}^{-1}$

Aij (nt) $n.10^{-2} \text{ s}^{-1}$

Fig. 7-4. Excitation rates for ammonia at 1 AU.